

nutrition

3 Types of malnutrition

1. **Marasmus:** lack somatic proteins
2. **Kwashiorkor:** lack visceral proteins
3. **Mixed Marasmus/Kwashiorkor:** a bit of both

3 Protein compartments

1. **Somatic:** muscle proteins
2. **Visceral:** produced by organs, e.g. albumin, transferrin
3. **Immunity:** a subset of visceral, e.g. antibodies, clotting factors

5 Steps to assess nutritional status

1. **Visual assessment**
 - Marasmus vs. Kwashiorkor
2. **Patient's history:** present and past weight
3. **Physical exam**
4. **Protein compartments**
 - Weight/Height measurements
 - Edema/obesity/pathologies distort results
 - Anthropometric measurements: muscle vs. fat amounts
 - Triceps skin fold
 - Mid-arm circumference
 - Creatinine/height index (CHI)
 - Urinary creatinine is decreased in malnutrition and renal disease
 - Better in terms of height than weight
 - CHI = actual/normal mg excreted (where 90%-100% is normal)
 - Albumin: 50-60% of total serum proteins
 - Normal conc. = 3.4-5g/100ml of blood
 - Catabolic stress = ↓albumin = ↑insulin = further ↓aa
 - Interplay between 2 types of malnutrition: somatic aa used as building blocks for visceral aa
 - Easier to detect drop in serum conc. of transferrin than albumin
5. **Immune competence** (*anergy*=inability of a patient to respond to an infection)
 - Lymphocyte counts
 - Antigen-skin tests: Multitest CMI system (think Cooties game)
 - Ability of lymphocytes and immunoglobulins to respond to antigens
 - >1 grade of swelling/redness indicates some immune response
 - 0 grade = lack of response = indicative of anergy and malnutrition

4 Roles of pharmacists in nutrition

1. Help diagnose type/degree of malnutrition
2. Recommend treatment/method
3. Evaluate treatment, suggest changes
4. Management

If the GI tract works, use it!

- Into mouth (**nasogastric tube**)
- Into stomach (**gastrostomy**)
- Into small intestine (**jejunostomy**)

If not, use parental/IV administration

- **Protein sparing**
 - Mild patients

- Peripheral arm vein
- Isotonic: 2.75-3.5% solution (+vitamins, minerals, electrolytes)
- Not a TPN modality: no fat or dextrose given
- **Peripheral TPN**
 - Mild to moderate patients for limited time (<10 days)
 - Patients with septicemia
 - No fluid restrictions
- **Central TPN**
 - Moderate to severe patients
 - Burn patients (no peripheral veins available)
 - Longer duration (2-3 weeks)
 - Subclavian or internal jugular vein

Numbers to know

Nitrogen balance = [Protein intake x 16%] – [N_{excreted} + C] → should have positive N_{balance} of 4-6g

Caloric conversions:

- Dextrose: 3.4kcal/g
- Lipid emulsion: 9kcal/g

Nutritional requirements:

- **Protein** (supplied in 3.5-15% conc.)
 - Maintenance 0.8-1.6g aa/kg/day
 - Mild trauma 1.6g-2.0g aa/kg/day
 - Severe trauma 2.0-3.0g aa/kg/day
- **Nitrogen**
 - 16% of protein -or- protein/6.25 (in grams)
- **Calories**
 - Mildly stressed 125-150kcal/gN₂
 - Highly stressed 80-100kcal/gN₂ (note: less calories needed for severe pts because less mobilized)
 - Provided as dextrose and fat (50/50, 40/60)
 - **Dextrose** supplied as 5-70% concentration (e.g. D₅W=50g/L, D₅₀W=500g/L)
 - **Lipid emulsion** supplied as 10-20% concentration
 - 200 extra kcal from glycerin and emulsifiers
 - Now favored over dextrose because
 - ↓Hyperglycemia risk
 - Dextrose → fat (accumulates in liver)
 - ↓Osmolarity of aa/dextrose solutions to facilitate peripheral admin
 - ↓Severity and frequency of phlebitis
 - ↓Chance of essential FA deficiency
- **Molecular weights**
 - Dextrose: 180g/mol
 - NS: 58.5g/mol

CALCULATIONS

Solve for nutrition

1. Solve for volume (ml) of **amino acids** needed [use: degree of trauma, weight of pt, concentration]
2. Solve for amount (g) of **nitrogen** needed [use: grams of amino acid from part 1]
3. Solve for **kcal** needed [use: degree of trauma, grams of nitrogen from part 2]
 - a. Determine **ratio** of dextrose/lipid

- Solve for volume (ml) of **dextrose** needed [use: ratio, concentration, caloric conversion]
- Solve for volume (ml) of **lipid emulsion** needed [use: ratio, concentration, caloric conversion]

Solve for tonicity

- Amino acids:** will be given mOsm/L, use known volume to solve
- Dextrose:** solve knowing D5W=280mOsm/L, multiply by known volume
- Lipid emulsion:** 2.25g glycerin/100ml...

Incompatibility charts

		ANIONS		
CATIONS		Monovalent Cl ⁻	Divalent CO ₃ ²⁻	Trivalent PO ₄ ³⁻
	Monovalent Na ⁺ , K ⁺	Soluble	Soluble	Soluble
	Divalent Ca ²⁺ , Mg ²⁺	Soluble	Insoluble	Insoluble
	Trivalent	Soluble	Insoluble	Insoluble

Mg²⁺ behaves differently → important when considering lactate ringers and ringer solutions

Ion pair incompatibilities

Mixing of a weak acid drug with a weak basic drug

Barbiturates: weak acids

Heparin: weak bases

Na+Phenobarb- + Morphine+SO₄- → big insoluble salt

Na+Phenobarb- + Tetracycline+Cl- → big insoluble salt

90% of reactions are some form of ion pair incompatibility

		LVP			
S V P		D5W + 5%NaHCO ₃ (basic)	0.9% NaCl (neutral)	D5RL (neutral, Ca ²⁺)	D5W + 15000 NaHeparin (basic)
	Dopamine HCl (basic)	I (free base)	C	C	I (ion pair precip)
	NaPhenytoin (acidic)	I (insoluble)	C	I (insoluble)	I (insoluble)
	KPenG (acidic)	I (base hydrolysis)	C	C	I (base hydrolysis)

SVP: have added sodium hydroxide (basic) but drug itself may be acidic

Penicillins: beta lactams or lactones, cyclic esters or amides

Beta lactam rings are sensitive to both acids and bases which cause hydrolysis

A lot of lactates are soluble: low enough molecular weight

NaPhenytoin is only soluble in NS and needs to be given w/in a ½ hr of making it up

NaPhenytoin is a weak acid but likes base, it is a very insoluble drug (in order to keep it ionized, need a lot of cosolvents and need to raise the pH really high)

		LVP		
S V P		0.9% NaCl (neutral)	D5RL (neutral)	D5W + 15000 NaHeparin (basic, w.acid drug)
	NaSucc of Hydrocortisone (w. acid)	C	I Ca ₃ (PO ₃) ₂	I (ester hydrolysis)
	Cimetidine HCl (w.base)	C	C	I (ion pair)
	Tetracycline HCl (w.base)	C	I (tet reacts w/ Ca ²⁺)	I (ion pair, free base)
	NaAmp + NaSucc of Hydrocortisone	C	I (amp inc	I (amp inc w/dextrose,

	(w.acid)		w/dextrose)	hydrolysis)	
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Benzyl alcohol: preservative

Sodium biphosphate anhydrous: buffer (phosphates are trivalent)

Ascorbic acid: antioxidant

Ampicillin can't be given in dextrose: it will hydrolyze the ampicillin

Hydrocortisone is an ester: ester hydrolysis

Tetracycline: don't take with milk or cheese because of it reacts with calcium

Barbituates: weak acid

Morphine sulfate: weak base

LVP				
S V P		RL +20000 NaHep (basic)	D20W + 4.25% aas (acid)	NS+50mg Hydrocortisone NaSucc ((acid)
	Lidocaine HCl (base)	I (ion pair, free base)	C	I (ion pair)
	NaKochalate (acid)	C	I (free acid)	C
	Haloperidol lactate ester	I (base hydrolysis of ester)	I (acid hydrolysis of ester)	C

When have big amounts of heparin → probably basic

If small amounts → not going to make it basic

In general, not good to put drugs in TPN

IMMUNE STIMULANTS

Vaccines

- Attenuated (live) or inactivated (killed)
- Whole or fractions
- Take virus → sonicate to death → differential centrifugation to isolate portion important for immune response

Toxoid

- Modified exotoxin rendered non-toxic
- Easy for companies to make
- Require refrigeration and/or reconstitution (freeze dried powder)

Passive immunity

- **Human immune sera**
 - Immunoglobulin: e.g. measles, hepatitis A
 - Hyper immune serum: e.g. hepatitis B, rabies
- **Animal immune sera**
 - Antitoxin: e.g. botulism, diphtheria, tetanus
 - Antiviral serum: e.g. rabies
 - Antivenin: e.g. rattle snake, black widow spider, scorpion
- **Viral vaccines** e.g. measles, mumps, influenza
 - Isolated by: disintegration, column filtration, differential centrifugation
- **Viral vaccines in human tissue culture**

Allergen extracts

- **Types:** food, animals, grasses, insects, molds, trees, weeds, inhalants
- **Extraction process:** percolation/decoction
- **Diagnosis:** scratch/prick test using very dilute solutions of allergen extracts

Immunotherapy

- Treatment of disease by inducing, enhancing, or suppressing an immune response
- Goal: to increase IgG
- Small subcut doses → weak dose qwk → increasing strength or # of doses over a few months
- IgG competes with IgE IgG>>>IgE with increased dosing with allergen
- Problems: difficult to standardize and establish potency
- Dosage forms: solutions, suspensions, lyophilized powders requiring reconstitution
- Additives
 - Human serum albumin: protein preservative
 - Normal saline: tonicity
 - Phenol: antimicrobial
 - Glycerin: protein preservative
 - Aluminum precipitated allergen: slows absorption/action of allergen
- Use & Handling
 - Aseptic and sterile procedures
 - Proper documentation
 - Generally require refrigeration (never frozen)
 - Sterility testing required (particulate/pyrogen testing not)

Hymenoptera venoms

- Honeybee, wasp, hornets
- Purified, lyophilized
- Usually only venom collected, not whole insect (exception: fire ants)

PROTEIN PHARMACEUTICALS

Biotechnological products techniques

- Recombinant DNA
- Monoclonal antibodies
- PCR
- Gene therapy
- Nucleotide blockade/antisense (mRNA)

Chemical instability

Change in structure due to breaking of bonds

- Proteolytic cleavage
 - Proteases
 - Hydrolysis: solution is to lyophilize the product
- Deamidation: cleave NH₃ group
- Oxidation: Met, Cys (also His, Trp, Tyr)

Physical instability

Change in structure *not* due to bond breaking/forming

- Aggregation: precipitation, normal Brownian motion
- Conformational stability: denaturing/unfolding of protein

Additives in biotech formulations

- **Serum albumin**
 - Flood the system so active proteins won't be bound, i.e. inhibit adsorption
 - Adsorption binding sites may be hydrophilic or lipophilic, proteins have both, which makes it a problem
 - Tubing: use polyester or nylon, not PVC, which has more binding sites
 - Resembles nascent complexing proteins: w/o nascent proteins, active proteins are more active
 - Cryoprotectant: protects while freeze drying, almost as good as mannitol
 - Examples when albumin is used: interferon, IL-2, TPA
- **Amino acids (Gly)**
 - Chelate trace elements to prevent aggregation : trace elements allow aggregation, e.g. Zn in insulin
 - Take up adsorption sites to reduce surface adsorption
 - Inhibit aggregate formation
 - Inhibit thermal induced inactivation: neutral amino acids like glycine protect the formulation if heated
- **Fatty acids & phospholipids**
 - 7-8 carbon length is optimal
 - Liposomal systems help stabilize proteins and peptides through nonpolar interactions
 - Protection of non-polar portion of protein interacting with non-polar lipid (emulsifiers for lipids are phospholipids)
- **Surfactants**
 - Charged surfactants (cationic/anionic) cause denaturation
 - Non-ionic surfactants stabilize by reducing interfacial tension
 - Reduce tendency for protein to unfold, help retain structure
 - Examples: Tween 80, Brij, poloxamer)
- **Metals**
 - Ca^{2+} and Cu^{3+} stabilize proteins
 - Bridge between disulfide bonds: helps stabilize bonds and tertiary structure
- **Polyols**
 - Polyhydroxyl groups: carbohydrates, sorbitol, mannitol, glycerol
 - Used in lyophilized dosage forms to prevent aggregation by adding bulk
 - Aids reconstitution: polyols are very water soluble, help the water get to the amino acid very quickly
 - Protects against oxidation
 - Strengthens intra hydrophobic bonds by reducing the interaction between water and protein
 - Humectant: helps hydrate the protein for protection and stabilization
- **Reducing agents**
 - Reduce disulfide bond formation, which tends to lower protein activity
 - Agents: glutathione, thioethanolamine, thiodiglycol, thioacetic acid, N-acetylcysteine
- **Chelating agents**
 - Problem: Cu, Fe, Ca, Mn act as catalysts in oxidation reactions by using up the oxygen
 - Chelating these metals help stabilize the formulation by using up the active sites on the metals
 - Agents: EDTA, diNa, CadiNa, tetraNa
- **Miscellaneous**
 - Hydrolyzed gelatin: available amino acids
 - Ammonium sulfate: adjusts pH